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Guide Learning goals

Checklist As you

prepare for your final

exam use the following

list to monitor your

progress. Only check

off a learning goal

when you are confident

of your ability to

complete the task

described. Unit 0:

Being a Chemist

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Recognize and apply safe practices when working in the lab.

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Chemistry Semester 2: Final Exam The Final Exam in this class is the Michigan Secondary Credit Examination for Chemistry. Exam Grades will be based on achieving the Department of

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recommended passing
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choice. MULTIPLE CHOICE Date (PRACTICE) You may use a pencil, eraser, and scientific calculator to complete the test. You will be given a periodic table. Useful formulas and the solubility rules are listed.

Chemistry Semester 2 Final Exam Multiple Choice

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Review Guide

In order to properly prepare for the final exam it is highly recommended that you complete this review guide in its entirety Chemistry semester 2 final answers. You should also create your cheat sheet during this time, including all of the information and formulas. .

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Multiple Choice and
Numerical Response 1.
Which of the following

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equations is associated with the largest energy change per mole of fluorine? A. $F_2(g) \rightarrow F_2(l)$ B. $19\ 9\ F + 4\ 2\ He \rightarrow 1\ 0\ n + 22\ 11\ Na$ C. $2\ F_2(g) + 2\ H_2O(l) \rightarrow O_2(g) + 4\ HF(aq)$

Final Exam Practice Test (Semester 2)

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NON-GRAPHING

SCIENTIFIC

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OF THE EXAM! You will

be provided with a

periodic table with the

polyatomic ion chart

and equations that

appear on the last

page of this packet.

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national exam is a
3-part, 4 hour and 45
minute exam
administered in mid or
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participate in this exam
each year.

**Chemistry Olympiad
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Candidate number:

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Examination Time

allowed: 09:00 - 18:00

(9 Hours) This

examination paper

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Section A (2 short

answer questions, each

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worth 10 marks for a total of 20 marks) and Section B (a descriptive summary, worth 15 marks). The maximum mark available is 35. There are a total of 3 pages including this cover page.

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