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a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6 \_\_\_\_ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c. group. b. series. d. sublevel block. \_\_\_\_ 22. In the modern periodic table, elements are ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original ...

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CHAPTER 5 REVIEW The Periodic Law MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53 a. What is its atomic number? 127 b. What is its mass number? atomic number c. Is the element's position in a modern periodic table determined by

### 5 The Periodic Law

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

### 3 Atoms: The Building Blocks of Matter

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### CH105: Chapter 1 - Measurements in Chemistry - Chemistry

Modern Chemistry 28 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 8. What is the frequency of a photon whose energy is  $3.4 \times 10^{-19}$  J? ( $h = 6.626 \times 10^{-34}$  J·s) a.  $8.8 \times 10^{26}$  Hz b.  $5.1 \times 10^{14}$  Hz c.  $1.9 \times 10^{-15}$  Hz d.  $2.3 \times 10^{-52}$  Hz \_\_\_\_ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's ...

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5. Identify the products when the following substances decompose: its separate elements metal oxide + water metal oxide + carbon dioxide water + sulfur dioxide a. a binary compound b. most metal hydroxides c. a metal carbonate d. the acid  $H_2SO_3$  co 6. The complete combustion of a hydrocarbon in excess oxygen yields the products 2 MODERN CHEMISTRY

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Chemistry is a branch of physical science that studies the composition, structure, properties and change of matter. Chemistry deals with such topics as the properties of individual atoms, how atoms form chemical bonds to create chemical compounds, the interactions of substances through inter-molecular forces that give matter its general properties, and the interactions between substances ...

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